

1718 measured reflections
1718 independent reflections

3 standard reflections
every 100 reflections
intensity decay: 2.6%

RefinementRefinement on F $R = 0.0448$ $wR = 0.0324$ $S = 1.48$

1341 reflections

135 parameters

Only H-atom U 's refined $w = 1/[\sigma^2(F) + 0.000225F^2]$ $(\Delta/\sigma)_{\max} = 0.015$ $\Delta\rho_{\max} = 0.97 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.99 \text{ e } \text{\AA}^{-3}$

Extinction correction: none

Scattering factors from *International Tables for X-ray Crystallography* (Vol. IV)

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Acta Cryst. (1997). **C53**, 217–219Table 1. *Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)*

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U^{ij} a_i^* a_j^* \cdot a_i \cdot a_j$$

	x	y	z	U_{eq}
Co1	0.5201 (1)	0.84488 (7)	0.8418 (2)	0.0214 (4)
Br2	0.5430 (1)	0.77576 (6)	1.1102 (2)	0.0359 (4)
Br3	0.2877 (1)	1.15424 (7)	0.9507 (2)	0.0494 (4)
Br4	0.5984 (1)	1.07330 (6)	0.6271 (2)	0.0388 (4)
N5	0.4519 (9)	0.7548 (4)	0.7220 (12)	0.028 (3)
N6	0.5676 (8)	0.9379 (4)	0.9672 (12)	0.024 (3)
N7	0.3191 (8)	0.8717 (4)	0.8905 (15)	0.035 (3)
N8	0.7225 (9)	0.8250 (4)	0.7958 (13)	0.031 (3)
N9	0.5217 (9)	0.8954 (4)	0.6124 (11)	0.030 (3)
C10	0.4385 (12)	0.9624 (6)	1.0671 (16)	0.041 (4)
C11	0.3100 (11)	0.9478 (6)	0.9555 (17)	0.037 (4)
C12	0.7740 (11)	0.8734 (6)	0.6507 (15)	0.035 (4)
C13	0.6530 (13)	0.8765 (6)	0.5188 (15)	0.039 (4)

Table 2. *Selected geometric parameters (\AA , $^\circ$)*

Co1—Br2	2.407 (2)	N6—C10	1.494 (14)
Co1—N5	1.976 (9)	N7—C11	1.468 (14)
Co1—N6	1.988 (9)	N8—C12	1.491 (15)
Co1—N7	1.974 (8)	N9—C13	1.459 (15)
Co1—N8	1.956 (9)	C10—C11	1.495 (16)
Co1—N9	1.972 (9)	C12—C13	1.514 (17)
Br2—Co1—N9	173.3 (3)	N7—Co1—N8	176.3 (4)
N5—Co1—N6	174.1 (4)	N8—Co1—N9	85.4 (4)
N6—Co1—N7	85.0 (4)		

The U_{iso} parameters of the H7A and H9A atoms were fixed to avoid unreasonable values. The enantiomeric crystal structure gave $R = 0.0547$ and $wR = 0.0413$, which are significantly greater than those of the present structure.

Structure analysis was carried out using *CRYSTAN-GM* software (Edwards, Gilmore, Mackay & Stewart, 1995) on a SUN SPARC10 workstation.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: TA1130). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Chelation of 2,6-Bis(hydroxymethyl)-pyridine in *cis*-Dichloro(triphenylphosphine-*P*)[2-(oxymethylene-*O*)-6-(hydroxymethyl)pyridine-*N*]oxorhenium(V)–Water (2/1)

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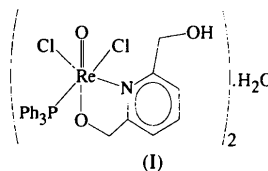
(Received 7 March 1996; accepted 17 October 1996)

Abstract

The potentially terdentate dianionic ligand pyridine-2,6-bis(methanol) (H_2L) acts as a bidentate monoanionic N, O^- -donor in the title compound, dichloro(oxo)[pyridine-2,6-bis(methanolato)(1-)- N, O](triphenylphosphine-*P*)rhenium(V)–water (2/1), $2[\text{ReCl}_2\text{O}\{\text{C}_5\text{H}_3\text{N}(\text{CH}_2\text{O})\text{-(CH}_2\text{OH)}\}(\text{PPh}_3)] \cdot \text{H}_2\text{O}$. The asymmetric unit contains two monomeric neutral independent and well separated $[\text{ReOCl}_2(\text{HL})(\text{PPh}_3)]$ moieties per water molecule.

Comment

As part of a program to study the formation of oxorhenium(V) complexes containing mixed di- and terdentate N, O^- -donor ligands (Du Preez, Gerber & Kemp, 1992; Gerber, Kemp, Du Preez & Bandoli, 1993; Bandoli, Gerber, Jacobs & Du Preez, 1994), we isolated the title compound, (I), in which the potentially terdentate 2,6-bis(hydroxymethyl)pyridine ligand (H_2L) acts as a bidentate N, O^- -donor.



The asymmetric unit contains two monomeric neutral independent and well separated [ReOC₂(HL)(PPh₃)] molecules (*A* and *B*) per water molecule (Fig. 1). There is no intermolecular bonding of significance and in both molecules the coordination geometry about the metal atom is distorted octahedral. Molecules *A* and *B* are well superimposable; in fact, the weighted r.m.s. deviation [derived from the *BMFIT* program (Nyburg, 1974)] is only 0.048 Å when the fitting is performed using the coordination geometry atoms. No bond distances and angles differ by more than two standard deviations, apart from the Re=O(1) distance which is 1.68(1) in molecule *A* and 1.64(1) Å in *B*. The Re atom is out of the mean equatorial plane [by 0.15(1) and 0.16(1) Å in *A* and *B*, respectively, towards the oxo O(1) atom] and the five-membered metalocycle assumes a twist-envelope conformation, with the Re atom out of the O(2)—C(6)—C(5)—N(1) ring by 0.09(1) and 0.10(1) Å in *A* and *B*, respectively. The structural properties (Table 2) are a common feature for numerous six-coordinate octahedral monooxo complexes of rhenium(V) and do not deserve any comment (Bandoli *et al.*, 1994, and references therein).

The water molecule engages in hydrogen bonding with the O(3A) atom at (*x*, *y*, *z* - 1) and no other interaction less than 3.3 Å can be observed in the packing.

Experimental

The synthesis of (I) was carried out by reaction of *trans*-[ReOC₂(PPh₃)₂] and C₅H₃N(CH₂OH)₂ (molar ratio 1:1) in an ethanolic solution for 1 h under refluxing conditions. Recrystallization was from acetonitrile.

Crystal data

2[ReCl₂O(C₇H₈NO₂)-
(C₁₈H₁₅P)].H₂O
M_r = 1365.0
Orthorhombic
*Pna*2₁
a = 16.331(6) Å
b = 29.61(1) Å
c = 10.854(5) Å
V = 5249(4) Å³
Z = 4
D_x = 1.727 Mg m⁻³
D_m not measured

Mo *K*α radiation
λ = 0.71073 Å
Cell parameters from 40
reflections
θ = 8–14°
μ = 4.923 mm⁻¹
T = 296 K
Parallelepiped
0.30 × 0.20 × 0.15 mm
Violet

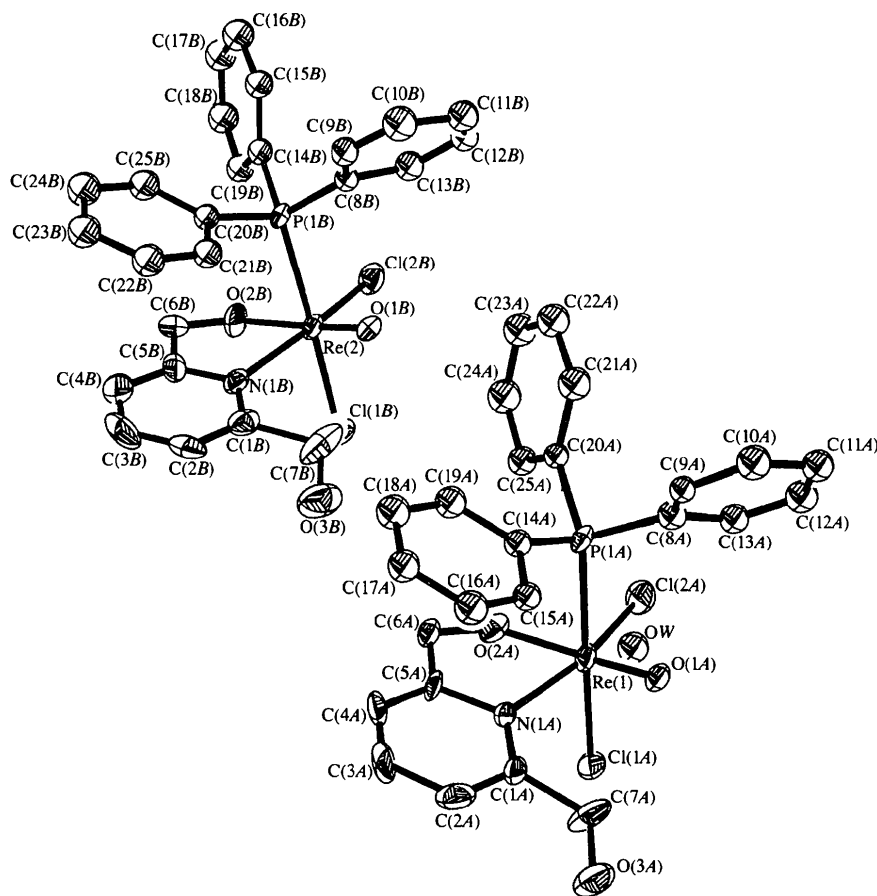


Fig. 1. The molecular structure of molecules *A* and *B* (ORTEP; Johnson, 1976) and the atom-labelling scheme. Displacement ellipsoids are drawn at the 40% probability level.

Data collection

Siemens R3m/V diffractometer	3741 reflections with $F > 4\sigma(F)$
$\omega/2\theta$ scans	$R_{\text{int}} = 0.0155$
Absorption correction: empirical (four ψ scans) (North, Phillips & Mathews, 1968)	$\theta_{\text{max}} = 25.0^\circ$
$T_{\text{min}} = 0.42$, $T_{\text{max}} = 0.48$	$h = 0 \rightarrow 19$
5162 measured reflections	$k = 0 \rightarrow 35$
4880 independent reflections	$l = 0 \rightarrow 12$
	2 standard reflections every 250 reflections
	intensity decay: none

Refinement

Refinement on F	$\Delta\rho_{\text{max}} = 1.87 \text{ e } \text{\AA}^{-3}$
$R = 0.0443$	$\Delta\rho_{\text{min}} = -1.67 \text{ e } \text{\AA}^{-3}$
$wR = 0.0597$	Extinction correction: none
$S = 0.85$	Scattering factors from <i>International Tables for X-ray Crystallography</i> (Vol. IV)
3741 reflections	Absolute configuration: Flack (1983)
419 parameters	Flack parameter = $-0.003(19)$
H atoms not refined	
$w = 1/[\sigma^2(F) + 0.0049F^2]$	
$(\Delta/\sigma)_{\text{max}} = 0.48$	

Table 1. Selected bond lengths (\AA) and angles ($^\circ$)

	Molecule A	Molecule B
Re—Cl(1)	2.418 (5)	2.428 (5)
Re—Cl(2)	2.355 (7)	2.345 (7)
Re—P(1)	2.468 (4)	2.461 (4)
Re—O(1)	1.680 (13)	1.639 (13)
Re—O(2)	1.923 (13)	1.862 (14)
Re—N(1)	2.205 (18)	2.206 (18)
Cl(1)—Re—Cl(2)	87.8 (3)	87.7 (3)
Cl(1)—Re—P(2)	177.8 (3)	177.4 (2)
Cl(1)—Re—O(1)	97.1 (4)	97.5 (5)
Cl(1)—Re—O(2)	94.4 (4)	92.3 (5)
Cl(1)—Re—N(1)	88.2 (5)	87.0 (5)
Cl(2)—Re—P(1)	89.9 (2)	91.9 (2)
Cl(2)—Re—O(1)	101.8 (6)	100.6 (5)
Cl(2)—Re—O(2)	90.2 (4)	92.0 (6)
Cl(2)—Re—N(1)	164.3 (5)	165.2 (5)
P(1)—Re—O(1)	83.4 (4)	85.1 (4)
P(1)—Re—O(2)	85.6 (4)	85.1 (5)
P(1)—Re—N(1)	93.9 (4)	92.8 (4)
O(1)—Re—O(2)	163.7 (6)	164.3 (7)
O(1)—Re—N(1)	93.8 (7)	93.9 (6)
O(2)—Re—N(1)	75.0 (6)	74.4 (7)

The H atoms, with the exception of the missing water H atoms, were included in calculated positions, but were not refined. Only a common variable displacement parameter was applied to them.

Data collection: *SHELXTL-Plus* (Sheldrick, 1987). Cell refinement: *SHELXTL-Plus*. Data reduction: *SHELXTL-Plus*. Program(s) used to solve structure: *SHELXTL-Plus*. Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993). Molecular graphics: *SHELXTL-Plus*.

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Lists of structure factors, anisotropic displacement parameters, atomic coordinates and complete geometry have been deposited with the IUCr (Reference: AB1373). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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A Tetranuclear Iron Cluster Complex: [$\{\text{Fe}(\text{CO})_3\}_3\{\mu_3\text{-CC}[\text{Fe}(\text{CO})_2(\eta^5\text{-C}_5\text{H}_5)\text{-C}(\text{CF}_3)]\}$]

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Abstract

Undecacarbonyl[4(η^5)-cyclopentadienyl]- μ_3 -(4,4,4-trifluorobutaneheptayl)-tetrairon(3 Fe—Fe), [$\{\text{Fe}(\text{CO})_3\}_3\text{-}\{\mu_3\text{-CC}[\text{Fe}(\text{CO})_2\text{Cp}]\text{C}(\text{CF}_3)\}$] (Cp is $\eta^5\text{-C}_5\text{H}_5$) contains a triangular $\{\text{Fe}(\text{CO})_3\}_3$ core [Fe—Fe distances 2.482 (1), 2.663 (1) and 2.693 (1) Å] linked *via* an unusual C_3CF_3 bridging unit to an $\text{Fe}(\text{CO})_2\text{Cp}$ group.

Comment

The dinuclear complex [$\{\text{Cp}(\text{CO})\text{Fe}\}_2\{\mu\text{-C}_2(\text{CF}_3)_2\text{-SMe}\}_2$] (Cp is $\eta^5\text{-C}_5\text{H}_5$), (1), reacts with tri-iron dodecacarbonyl [$\text{Fe}_3(\text{CO})_{12}$] to give $\mu_3\text{-}\eta^2$ -alkyne